

PHTHALAZINE DERIVATIVES FROM AROMATIC ALDAZINES

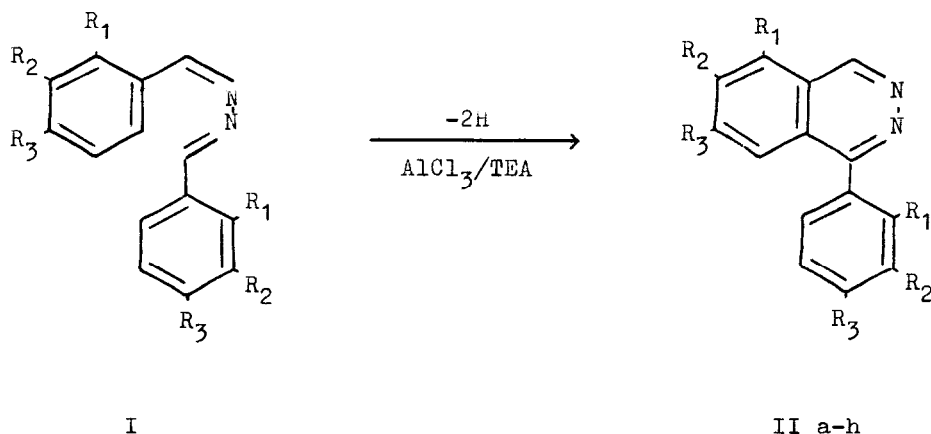
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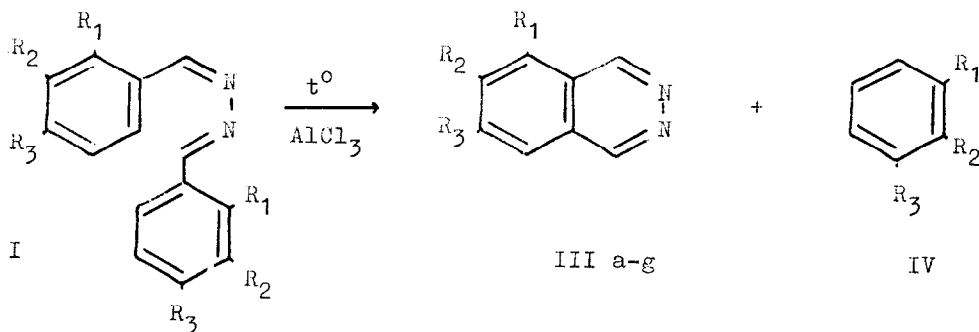
Summary: A Lewis acid mediated synthesis of phthalazine derivatives II and III from aromatic aldazines I is reported.

In the present paper we report a new synthesis of 1-arylphthalazine derivatives II from aromatic aldazines I by heating with aluminium chloride/triethylamine (TEA) (molar ratio 2:1) at 170-200°C.



$AlBr_3/TEA$ also proved to be a good reagent, promoting formation of II from I, the yield being up to 70%. We have found that when $AlCl_3$ alone reacts with I (molar ratio $AlCl_3:I$ more than 12) mainly elimination of one of the aryl residues occurred as arene or haloarene IV and phthalazine derivatives III are for

med unsubstituted in the pyridazine nucleus. 3-Chloro- and 3-methyl- substituted benzaldazines gave mixtures of the corresponding 5- and 6-substituted phthalazines, the ratio being approximately 5:1. From 3-bromobenzaldazine we isolated 6-bromophthalazine.

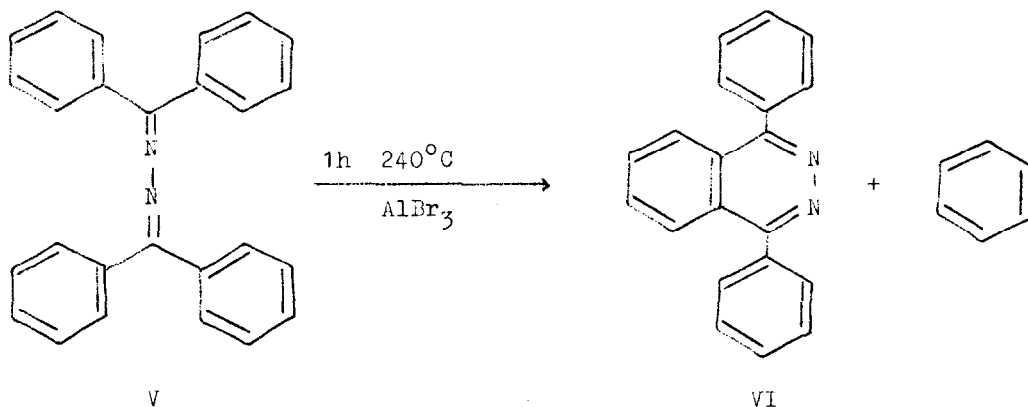


pure AlBr_3 , compared with AlCl_3 and AlX_3/TEA , occupies intermediate position with regard to the cyclization of I, giving rise both to II and III. In the synthesis of III its addition to AlCl_3 is useful as "liquefier" in the temperature range $170\text{--}210^\circ\text{C}$.

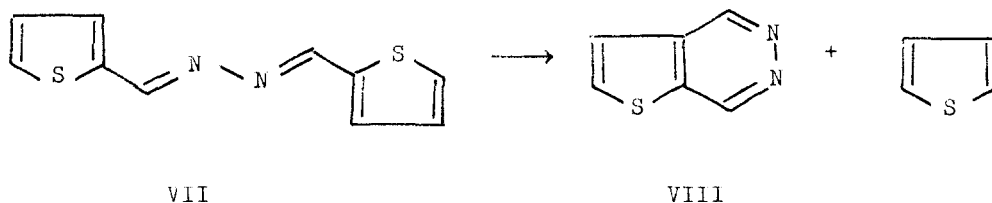
II	R ₁	R ₂	R ₃	mp	t ^o (yield %)	III	R ₁	R ₂	R ₃	mp	t ^o (yield %)
a	H	H	H	149	(1) 63	a	H	H	H	91	(2) 43
b	H	H	CH ₃	102	68	b	H	H	CH ₃	72	18
c	CH ₃	H	CH ₃	114	60	c	CH ₃	H	CH ₃	106	65
d	H	H	Cl	193	70	d	H	H	Cl	139	20
e	F	H	H	107	27	e	H	H	Br	110	(3) 35
f	H	H	iPr	132	40	f	Cl	H	H	128	45
g	H	H	Br	208	65	g	H	H	C ₆ H ₅	139	(4) 15
h	Cl	H	H	156	62						

The elimination of IV during the synthesis of III seems to be of wider validity since we have further established that a similar process takes place also in the case of benzophenone azine V from which 1,4-diphenylphthalazine VI⁽⁵⁾ was synthesised in 60 % yield, with liberation of benzene. All compounds of types II and III are newly synthesised, except IIa, IIIa, IIIe and IIIg. Their

structure was elucidated on the basis of elemental microanalysis, nmr, ir, UV and mass spectral data.



In analogous way thieno/2,3-d/pyridazine VIII was synthesised from thiophene-2-aldehyde azine VII in 40% yield, which might be a synthesis of choice if the necessary thiophene-2,3-dialdehyde used by Robba et al. for the synthesis of the same compound⁽⁶⁾ is not at hand.



It should be pointed out however that we were not able to obtain the expected pyrido/2,3-d/pyridazine when the reaction was tried with pyridine-2-aldehyde and the starting aldazine was quantitatively recovered.

1-(2,4-Dimethylphenyl)-5,7-dimethylphthalazine IIc

0,53g (2 mmole) 2,4-Dimethylbenzalazine Ic and 2,0g (15 mmole) $AlCl_3$ were mixed together and 1,0 ml (7,17 mmole) TEA was added, with cooling. After 1 hour at 180-200°C the dark homogeneous reaction mixture was cooled and treated with 30 ml ice water. The resulted suspension was filtered and washed with

dilute HCl and the combined filtrate was made alkaline with 15% KOH. The precipitate was collected, dried and recrystallised from n-hexane giving IIc (0,31g, 60%, mp 113-114°C).

The aqueous filtrate after the separation of IIc was extracted with benzene and the solvent distilled off under reduced pressure. The residue gave a negligible amount of crystals (mp 105°C) which were identified as 5,7-dimethylphthalazine IIIc.

$C_{18}H_{18}N_2$, M^+ 262; UV(ethanol) λ max(log e) nm 231(4,83), 287(3,97);
 1H nmr, 80 MHz, δ $\overset{TMS}{CDCl_3}$ 2,13(3H,s, o-CH₃), 2,37(6H,s, p-CH₃, 5-CH₃), 2,80(3H,s, 7-CH₃), 6,90-7,60(5H,m, arom.), 9,65(1H,s, H₄) ppm. Picrate mp 164°C (ethanol).

5,7-Dimethylphthalazine IIIc 0,53g (2 mmole) Ic, 4,0g (30 mmole) AlCl₃ and 4,0g (15 mmole) AlBr₃ were heated at 185-195°C for 30 min, the reaction mixture treated under cooling with 50 ml water and after addition of 5 ml 5% HCl the filtered solution was made alkaline with 15% KOH and extracted thoroughly with benzene. After evaporation of the organic solvent the crude IIIc crystallised in needles which recrystallised from benzene/n-hexane melted at 105-106°C; yield 0,21g(65%).

$C_{10}H_{10}N_2$, M^+ 158; UV(ethanol) λ max(log e) nm 228(4,85), 276(3,86);
 1H nmr, 80 MHz, δ $\overset{TMS}{CDCl_3}$ 2,60(3H,s, 5-CH₃), 2,75(3H,s, 7-CH₃), 7,50(2H, H₆ & H₈), 9,38(1H,s, H₄), 9,58 (1H,s, H₁) ppm. Picrate mp 208°C(ethanol).

REFERENCES

1. S.Gabriel, G.Pinkas, Ber.Deut.Chem.Ges., 23(1893)2210.
2. A.Lieck, Ber.Deut.Chem.Ges., 38(1905)3918.
3. J.H.Bowie, R.G.Cooks, P.F.Donaghue, J.A.Hallday, h.J.Rodda, Austr.J.Chem., 20(1967)2677.
4. C.M.Atkinson, C.J.Sharpe, J.Chem.Soc., (1959)2858.
5. A.Mustafa, A.H.Harhash, A.A.S.Saleh, J.Am.Chem.Soc., 82(1960)2738.
6. M.F.Robba, B.P.Roques, Fr.Pat. 1,453,897(Cl.C 07d)(1966), Chem.Abstr., 66(1967) P 95066a.

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